

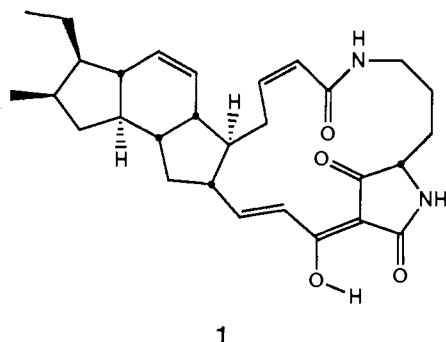
An Enantioselective and Highly Convergent Synthesis of (+)-Ikarugamycin

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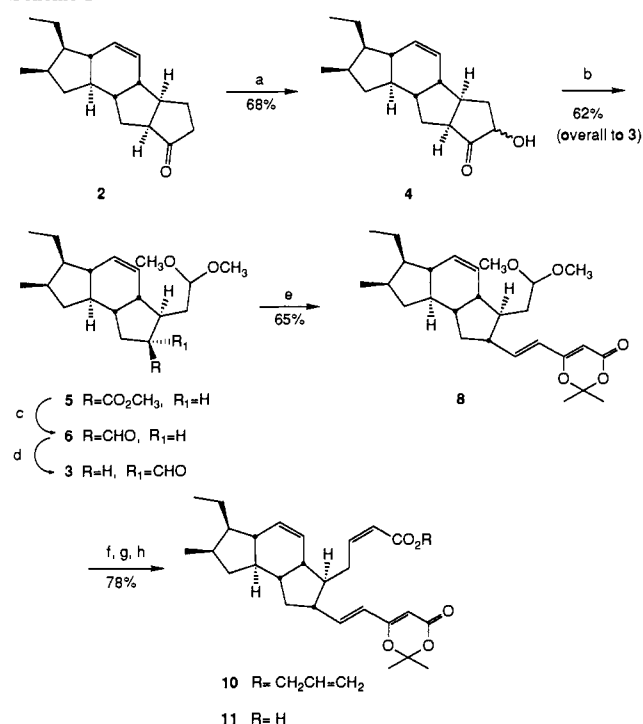
(+)-Ikarugamycin (**1**), the initial member of a novel class of macrocyclic lactam tetramic acid antibiotics possessing a range of biological activity, was isolated in 1972.¹ The structure and absolute stereochemistry of **1** was assigned in 1976 employing a combination of chemical degradation and spectroscopic methods.² A retrosynthetic analysis of **1** led to the identification of tetracyclic ketone **2** as a key intermediate. A convergent, highly stereoselective synthesis of (+)-**2** was developed in our laboratories employing a key intramolecular Diels-Alder reaction to assemble the required *as*-hydrindacene nucleus.³ An efficient, convergent solution to the formidable challenge presented by the tetramic acid-containing macrocyclic lactam subunit of **1** was also devised in our laboratories utilizing a new ketene-mediated cyclization for the construction of the 16-membered macrocyclic lactam ring.⁴ Herein we describe the application of this strategy to the total synthesis of (+)-**1**.⁵



In order to implement the aforementioned sequence to (+)-**1** beginning with (+)-**2**, elaboration of **2** to a differentially protected tricyclic dialdehyde **3** was required to allow the sequential construction of the *E* and *Z* unsaturated carbonyl residues present in **1** (Scheme I). With suitably differentiated appendages in place, coupling of the *Z* unsaturated carbonyl function with an appropriately protected L-ornithine derivative, followed by macrocyclization and formation of the tetramic acid, then completes the conversion to **1**.

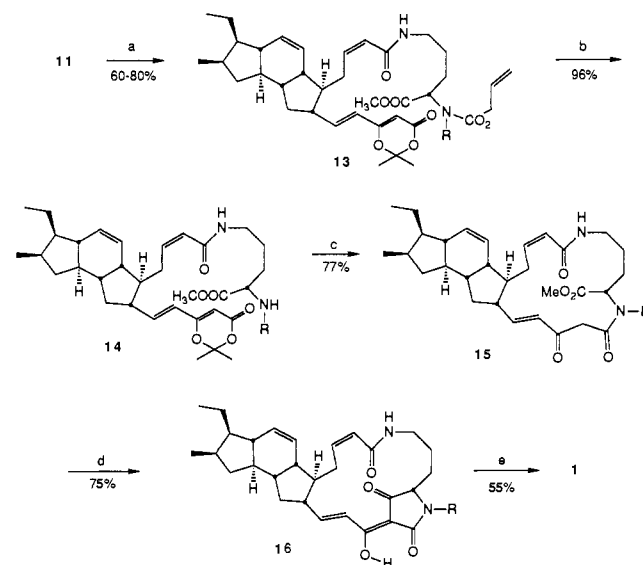
Elaboration of (+)-**2** (mp 65–67 °C, $[\alpha]_D^{25} + 103^\circ$ (*c* 2.8, CHCl₃)) was initiated by oxidation with PhI(OAc)₂ to afford a mixture of acyloins **4** (4:1 α/β) in 68% yield (Scheme I).^{6,7} Subsequent oxidative cleavage of **4** with Pb(OAc)₄ in anhydrous CH₃OH and immediate protection of the resulting ester aldehyde

Scheme I^a



^a Reagents: (a) PhI(OAc)₂ (1.2 equiv), KOH (2 equiv), CH₃OH, 25 °C, 8 h then Amberlyst-15, THF–H₂O (95:5 (v/v)), 25 °C, 24 h; (b) Pb(OAc)₄ (1.05 equiv), CH₃OH–THF (1:1 (v/v)), 0 °C, 0.5 h then Amberlyst-15, 3 Å molecular sieves, CH₃OH, 25 °C, 16 h; (c) DiBAL-H (2 equiv), THF, 0 °C, 2 h; (d) PDC (2 equiv), 3 Å molecular sieves, CH₂Cl₂, 25 °C, 0.5 h, then DBU (catalytic), CH₂Cl₂, 0 °C, 72–120 h; (e) **7** (1.2 equiv), KHMDS (1.2 equiv), THF, 0 °C → 25 °C, 4 h; (f) Amberlyst-15, CH₃CN–H₂O (9:1 (v/v)), 25 °C, 12 h; (g) **9** (1 equiv), K₂CO₃ (6 equiv), 18-c-6 (10 equiv), PhCH₃, –20 °C → 0 °C, 4 h; (h) NH₄OAc (4 equiv), Pd(PPh₃)₄ (catalytic), dioxane, 25 °C, 24 h.

Scheme II^a



^a Reagents: (a) mesitylene sulfonyl chloride (1 equiv), Et₃N (1 equiv), THF, 25 °C, 10 min then **12** (2–3 equiv), DMAP (3–4 equiv), THF, 25 °C, 4 h; (b) HOAc (xs), Pd(PPh₃)₄ (catalytic), THF, 25 °C, 12 h; (c) PhCH₃, 105 °C, 8–10 h; (d) *t*-BuOK (2 equiv), *t*-BuOH, 0 °C, 15 min; (e) anhydrous TFA (0.01 M in substrate), 72 °C, 5 min.

cleanly provided ester acetal **5** (90% from **4**). Utilizing standard chemistry, **5** could be converted to aldehyde acetal **6**, which as expected underwent epimerization to the more stable trans aldehyde **3** on exposure to DBU (62% overall from **4**).^{8,9} Installation

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(3) Boeckman, R. K., Jr.; Napier, J. J.; Thomas, E. W.; Sato, R. I. *J. Org. Chem.* **1983**, *48*, 4152. For other approaches to the carbocyclic system of **1**, see: Whitesell, J. K.; Minton, M. A. *J. Am. Chem. Soc.* **1987**, *109*, 6403 and references therein.

(4) For a discussion of the various synthetic strategies for formation of the macrocyclic ring, see: Boeckman, R. K., Jr.; Perni, R. B. *J. Org. Chem.* **1986**, *51*, 5486.

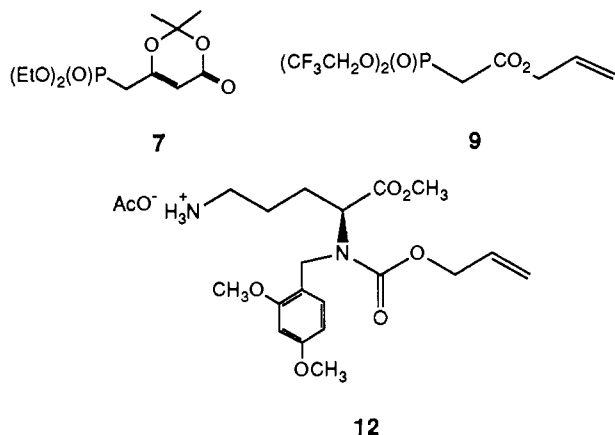
(5) See the following paper in this issue for an alternative total synthesis of (+)-**1** by Paquette and co-workers: Paquette, L. A.; Macdonald, D.; Anderson, L.; Wright, J. *J. Am. Chem. Soc.* **1989**, *111*, following paper in this issue.

(6) All new substances exhibited satisfactory spectroscopic (NMR, IR, UV) and combustion or high-resolution mass spectral analytical data.

(7) Moriarity, R. M.; Hou, K. C. *Tetrahedron Lett.* **1984**, *25*, 691. Use of highly electrophilic oxidants was precluded by the unexpectedly high reactivity of the strained double bond in **2**.

of the *trans* unsaturated ketene precursor was then effected by Horner-Emmons reaction of **3** with dioxinone phosphonate **7** which afforded exclusively the required *E* unsaturated dioxinone **8** in 65% yield.¹⁰

The required *Z* olefinic side chain was then elaborated via a *cis* selective Horner-Emmons reaction¹¹ of the aldehyde obtained by mild hydrolysis of **8**¹² with allyl bis-trifluoroethylphosphonoacetate **9** providing the *Z* allyl ester **10** (19:1 *Z/E*).¹³ Deprotection of allyl ester **10** with Pd(PPh₃)₄ (catalytic) and NH₄OAc then afforded acid **11** with no detectable double bond isomerization (78% overall from **8**).¹⁴



The crucial coupling of acid **11** and the primary amine derived from ammonium salt **12**,¹⁵ which proved to be remarkably sensitive to reaction conditions, was realized via addition of **12** to the mixed mesitylene sulfonic anhydride derived from **11** and subsequent treatment with DMAP providing allyl carbamate **13** (60–80% yield).^{16,17} Deblocking of **13** (Pd(PPh₃)₄ (catalytic)/HOAc) to secondary amine **14** (95%) has now set the stage for formation of the macrocyclic lactam.¹⁴

As hoped, heating a dilute solution of **14** in PhCH₃ (10⁻²–10⁻⁴ M) at 105 °C for 8–10 h cleanly provided the macrocyclic bis-amide **15** (~80% yield) via intramolecular trapping of the resulting highly electrophilic acyl ketene.¹⁸

Transannular Dieckmann cyclization of the highly constrained bis-amide **15** proceeded with noteworthy facility employing standard conditions (*t*-BuOK/*t*-BuOH) affording the penultimate intermediate *N*-(2,4-dimethoxybenzyl)ikarugamycin (**16**) in 75%

yield.¹⁹ Deprotection of **16** was then achieved by brief heating of a solution of **16** in anhydrous TFA (0.01 M) at 72 °C providing synthetic (+)-ikarugamycin (**1**) in 55% yield which was chromatographically and spectroscopically indistinguishable from natural (+)-**1**.^{20–22}

The foregoing total synthesis confirms both the structure and absolute stereochemistry previously assigned to (+)-ikarugamycin (**1**) by Ito and Hirata.² The sequence for conversion of (+)-**2** to (+)-ikarugamycin (**1**) proceeds in about 12 steps. Overall, (+)-ikarugamycin (**1**) is available in about 28 steps (longest linear sequence) from L-glyceraldehyde acetoneide.

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Supplementary Material Available: ¹H NMR spectra for compounds **1** (synthetic and natural) and intermediates **2–6**, **8**, and **10–16** (15 pages). Ordering information is given on any current masthead page.

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(21) Natural (+)-**1**: mp 228–229 °C (CH₃OH), λ_{max} 229, 323 nm, [α]_D²⁵ + 360° (c 0.19 DMF), α_D²⁵ + 289° (c 0.12, THF), R_f (silicic acid (Biosil A)) CHCl₃–CH₃OH (9:1), 0.3–0.5 R_f (SiO₂-G, (E. Merck)) EtOH–CH₃CO₂H (4:1), 0.56.² Synthetic (+)-**1**: mp 224–226 °C (CH₃OH), mmp 224–226 °C, λ_{max} 229, 323 nm, [α]_D²³ + 271° (c 0.10, THF), R_f (silicic acid (Biosil A)) CHCl₃–CH₃OH (9:1), 0.3–0.5, R_f (SiO₂-G (E. Merck)) EtOH–CH₃CO₂H (4:1), 0.56.

(22) We thank Fujisawa Pharmaceutical Co. Ltd., Higashiyodogawa-ku, Osaka, Japan for an authentic sample of natural (+)-ikarugamycin (**1**) for comparison with synthetic material.

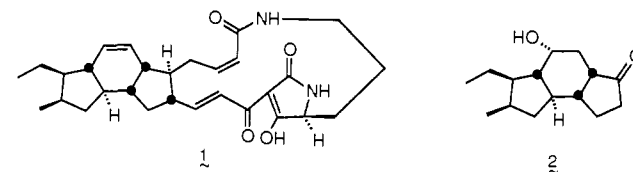
A Triply Convergent Enantioselective Total Synthesis of (+)-Ikarugamycin

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The isolation in 1972 of (+)-ikarugamycin (**1**),² an antibiotic possessing antiprotozoal, antiamebic, and gram-positive activity, was followed quickly by its characterization as a structurally unusual macrocyclic lactam embodying both an enoyltetramic acid



moiety and a *trans,anti,cis*-decahydro-*as*-indacene subunit.³ The challenge surrounding construction and proper amalgamation of

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(13) A variant of the Still procedure¹¹ was employed to prepare **9** (35% overall yield) from ethyl diethylphosphonoacetate: (a) KOH (1 equiv), CH₃CH₂OH, 25 °C, 16 h; (b) CH₂=CHCH₂Br (2 equiv), DMF, 25 °C, 24 h; (c) PCl₅ (2.2 equiv), 75 °C, 3 h; CF₃CH₂OH (2.1 equiv), (*i*-Pr)₂EtN (2.1 equiv), PhH, 25 °C, 12 h.

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(15) Ammonium salt **12** was synthesized from L-ornithine·HCl (29% overall yield) via the five-step sequence: (a) CuCO₃ (1.4 equiv), H₂O, 100 °C, 1 h followed by Cl₃CC(CH₃)₂OCOC(1.2 equiv) NaHCO₃, H₂O, 25 °C, 24 h then H₂S(g); (b) *t*-BuOCO₂N=C(C₆H₅)CN (1.2 equiv), Et₃N (1.5 equiv), dioxane–H₂O (1:1 (v/v)), 25 °C, 24 h; (c) CH₂N₂ (1.1 equiv), Et₂O, 0 °C, 0.5 h; (d) TFA–CH₂Cl₂ (1:5 (v/v)), 0 °C, 1 h; (e) 2,4-(CH₃O)₂PhCHO (1 equiv), PhCH₃, then evaporate (<0.5 mm) followed by NaCNBH₃ (2 equiv), CH₃OH, 25 °C, 1 h; (f) CH₂=CHCH₂OCOC(2 equiv), DMAP (2 equiv), ClCH₂CH₂Cl, 70 °C, 24 h; (g) Zn (10 equiv), HOAc, 25 °C, 2 h.

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